

Anti-inflammatory and anticancer activities of (–)-zeylenol from stems of *Uvaria grandiflora*

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ABSTRACT: *Uvaria grandiflora* has stimulated rigorous phytochemical investigation aimed at determining chemical structures and biological activities. In recent years, many polyoxygenated cyclohexenes have been isolated from this genus. This study reports that zeylenol isolated from the stems of *U. grandiflora* possess potential anti-inflammatory, anticancer, and caspase-3 activities. The structure of this compound was elucidated from spectroscopic analysis, particularly 2D NMR techniques. The anti-inflammatory effect and toxicity of zeylenol were evaluated in animal models and compared to that of reference drugs. In addition, the cytotoxicity of the isolated compound against human breast cancer MDA-MB231 and hepatocellular carcinoma HepG2 cell lines were studied. The test (–)-zeylenol at the dose of 1 mg/ear significantly inhibited oedema formation by 90%, 69%, 52%, and 52% after 15, 30, 60, and 120 min, respectively. Zeylenol was found to be toxic to both MDA-MB231 and HepG2 cells in a dose response manner with IC₅₀ values of 54 ± 10 and > 80 μM, respectively. The compound induced MDA-MB231 cell apoptosis via caspase-3 activation. Zeylenol probably possesses anti-inflammatory activity by inhibiting the synthesis or release of various inflammatory mediators and might be used to induce human breast cancer MDA-MB231 cell apoptosis.

KEYWORDS: cytotoxicity, MDA-MB231, HepG2, caspase-3 activity, Annonaceae

INTRODUCTION

The genus *Uvaria*, one of the largest palaeotropical genera in the family Annonaceae, comprises more than 220 species distributed in wet tropical regions of Africa, Madagascar, southeast Asia, northern Australia, and Melanesia. In Thailand, *U. grandiflora* is found in the south and southeast regions. Plants in this genus have been studied for bioactive constituents and various classes of compounds including alkaloids, annonaceous acetogenins, flavonoids have been isolated. Several compounds from this genus show antimalarial, antitumour, pesticidal and other biological activities^{1–4}.

U. grandiflora is a hardwood forest vine, large, and sticky. A single flower at the end of twig reaches 6–12 cm diameter when blooming. Phytochemical investigation of *U. grandiflora* demonstrated the presence of several groups of natural chemical, including polyoxygenated cyclohexenes and aromatic derivatives, also showed interesting antitumour, antimalarial^{1–4}, and antipancreatic cancer activities^{5,6}. Polyoxygenated cyclohexenes have a rather simple molecular skeleton of 1-methylcyclohex-4-ene. However, they contain multiple oxygenic substituents, such as benzyloxy, hydroxyl, alkoxy, epoxy, and acetoxyl groups with abundant stereoisomers^{7–9}. Previously, zeylenol has been reported to occur in two configura-

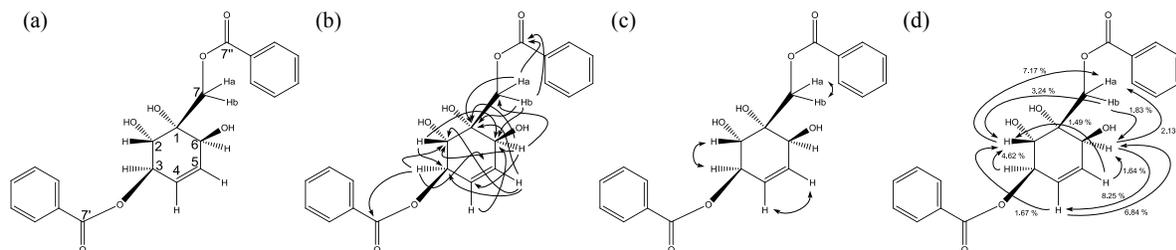


Fig. 1 (a) Structure of zeyleenol together with significant correlations in the (b) HMBC, (c) COSY and (d) NOE spectra.

tions as (–)-zeyleenol and (+)-zeyleenol from various species of the *U. genus* and other genera^{10,11}. In the current study, we reported the isolation, structural elucidation, anti-inflammatory and anticancer evaluation of (–)-zeyleenol (Fig. 1), a polyoxygenated cyclohexene derivative which was isolated from an ethyl acetate extract of stems of *U. grandiflora*.

MATERIALS AND METHODS

General experimental procedures

Column chromatography (CC) was carried out over silica gel (70–230 mesh, MERCK). Fractions obtained from CC were monitored by TLC (pre-coated silica gel 60 F₂₅₄, 20 × 20 cm, MERCK). UV spectrum was obtained on a Shimadzu UV-1601 spectrophotometer with ethanol as solvent. Melting point was measured on a Büchi 322 micro melting point apparatus and was uncorrected. Optical rotation was determined with a JASCO DIP-370 digital polarimeter. IR spectrum (KBr disk) was recorded on Shimadzu 8900 FTIR spectrophotometer. High resolution mass spectrum (HRESIMS) was measured on a micromass Q-TOF-2 (Waters) spectrometer. Low resolution mass spectrum was recorded on a Thermo Finnigan Polaris Q mass spectrometer. NMR spectroscopic data were obtained from a Bruker DPX 400 spectrometer. The chemical shifts were recorded in δ values and referenced to tetramethylsilane as the internal standard.

Plants material

The stems of *U. grandiflora* were collected from Trang Province of Thailand in October 2010 and identified by Narong Nuntasana. A voucher specimen (BKF 132441) was deposited at Forest Herbarium, Department of National Park, Wildlife and Plant Conservation, Ministry of Natural Resources and Environment, Bangkok, Thailand.

Phytochemical characterization

Air-dried powders of stems of *U. grandiflora* (2.2 kg) were extracted successively at room temperature with hexane, ethyl acetate, and methanol (3 × 5 l for each

solvent). Further separations of ethyl acetate extract (56.63 g) by silica gel CC was carried out, eluted with a gradient system between hexane and ethyl acetate to afford eight fractions. The fraction six was rechromatographed by the same procedure to yield five subfractions. The precipitate in the subfraction three was recrystallized from ethanol to obtain the isolated compound as a white solid (300 mg). Based on the comparison of its spectrometric data, e.g., MS, IR, UV, ¹H, ¹³C NMR together with 2D NMR (HMBC, COSY and NOE) experiments to those reported in the literature, compound was identified to be (–)-zeyleenol¹⁰ (Fig. 1).

(–)-Zeyleenol as an isolated compound had the molecular formula C₂₁H₂₀O₇, deduced from the HRESIMS (found m/z 385.1300 [M + H]⁺). The UV spectrum showed maxima at 274 and 231 nm, suggesting an extended benzoyl chromophore. The IR spectrum showed the presence of hydroxyl groups (strong absorption at 3469 cm⁻¹), ester groups (strong absorption at 1693 cm⁻¹), and monosubstituted phenyl rings (1600, 1585, and strong absorption 711 cm⁻¹). The EIMS at m/z 105, benzylum ion of zeyleenol formed by the cleavage of the ester groups. Moreover, a peak at m/z 77 formed by competitive cleavage of the C-1'', C-7'' or C-1', C-7' also supported the presence of phenyl groups in the structure.

The presence of two benzoyl groups was confirmed by the aromatic proton signals δ 7.32–7.96 (10H, dd). The typical methylene protons bearing benzoyloxy group appeared at δ 4.86 (d, $J = 12.3$ Hz, 1H) and δ 4.70 (d, $J = 12.3$ Hz, 1H). The ¹H NMR spectrum also showed two olefinic protons at δ 5.98 (ddd, $J = 10.2, 3.8, 1.6$ Hz, 1H) and δ 5.82 (dd, $J = 10.2, 2.9$ Hz, 1H), three carbonyl protons at δ 4.38 (m, 1H), δ 4.27 (t, $J = 4.4$ Hz, 1H) and δ 5.70 (m, 1H). The key HMBC correlations from methine proton H-6 (δ 4.38) to C-4 (δ 126.5) and H-3 (δ 5.70) to C-5 (δ 130.1) supported that the double bond located at C-4/C-5. The HMBC correlations of H-7a and H-7b to the ester carbonyl at δ_c 167.7 and H-3 to ester carbonyl δ_c 167.1 indicated that the two

benzoxyl groups located at C-1 and C-3, respectively. Further, in the ^{13}C NMR spectrum, one carbon of tertiary alcohol resonated at δ 75.9 (C-1) and one carbon of secondary alcohol resonated at δ 70.7 (C-2). In the COSY spectrum, a strongly proton triplet at δ 4.27 (H-2) showed coupling to methine proton absorption at δ 5.7 (H-3) and olefinic proton δ 5.82 (H-4) with proton δ 5.98 (H-5) and methylene proton H-7a (δ 4.86) with proton H-7b (δ 5.70), respectively.

Furthermore, on the basis of 1D NOE NMR experiment of zeyleanol, it was established that the hydroxyl group (C-2) and benzoxyl group (C-3) were both axial, i.e., β -configuration. Moreover, the relative configuration between H-2 and H-7a of benzoxyl moiety was also established as *cis*-configuration. This structure corresponds to zeyleanol as reported by Pan et al⁷ and Jolad et al¹⁰.

Zeyleanol: White needle crystals, m.p. 110.1–112.2 °C; $[\alpha]_{\text{D}}^{25}$ -41.16° ($c = 0.26$, CHCl_3); UV λ_{max} (EtOH) nm ($\log \epsilon$): 274 (4.66) and 231 (5.46); $^1\text{H-NMR}$ (CDCl_3) δ : 7.96–7.32 (5H, dd, $J = 8.2$, 0.8 Hz), 7.91–7.34 (5H, dd, $J = 8.2$, 0.8 Hz), 5.98 (1H, ddd, $J = 10.2$, 3.8, 1.6 Hz), 5.82 (1H, dd, $J = 10.2$, 2.5 Hz), 5.70 (1H, m), 4.86 (1H, d, $J = 12.3$ Hz), 4.70 (1H, d, $J = 12.3$ Hz), 4.38 (1H, br.t, $J = 3.9$ Hz), 4.27 (1H, t, $J = 4.9$ Hz). ^{13}C NMR (CDCl_3) δ : 167.7 (s), 167.1 (s), 133.4 (d), 133.3 (d), 130.1 (d), 129.8 (s), 129.7 (d), 129.4 (d), 129.3 (s), 128.4 (d), 128.3 (d), 126.5 (d), 75.9 (s), 73.9 (d), 70.7 (d), 68.8 (d), 66.6 (t). IR (KBr) cm^{-1} : 3469, 1693, 1600, 1585, 1452, 1317, 1282, 1120, 1072, 711. EI-MS m/z : 384(0.32), 215(6), 202(2), 131(1), 121(8), 105(100), 77(70), 50(20). HRESIMS (found m/z 385.1300 $[\text{M} + \text{H}]^+$).

Animals

Male Sprague-Dawley rats weighing 40–60 g purchased from the National Laboratory Animal Centre, Nakorn Pathom Province, Thailand, were used. All animals were kept in a room maintained under environmentally controlled conditions of $24 \pm 1^\circ\text{C}$ and 12 h light 12 h dark cycle. The animals had free access to water and food. They were acclimatized at least 1 week before starting the experiments. The experimental procedure was approved by Animal Ethics Committees, Faculty of Medicine, Chiang Mai University, Thailand. (Internal protocol number 26/2551, approved on 5 Feb 2009).

Anti-inflammatory effect

Oedema was assessed with the method of Brattsand et al¹². Male rats weighing 40–60 g were used. Ear oedema was induced by the topical application of

either EPP dissolved in acetone to the inner and outer surfaces of both ears by means of an automatic microlitre pipette. Test zeyleanol, at the dose of 1 mg/ear, were dissolved in acetone and applied topically in a volume of 20 μl to the inner and outer surfaces of the pinna just before the irritants. The control group received acetone.

Cell culture

The human hepatocellular carcinoma HepG2 and breast cancer MDA-MB231 cells were cultured in Leibovitz's L-15 and Dulbecco's modified eagle media (DMEM), respectively, with 25 mM NaHCO_3 , 20 mM HEPES, 100 units/ml penicillin, 100 $\mu\text{g}/\text{ml}$ streptomycin and supplemented with 10% foetal bovine serum. The cell lines were grown at 37°C in a 5% CO_2 atmosphere. The cells (1×10^3) were treated with the purified compounds at indicated concentrations for 24 h. The (–)-zeyleanol compound was dissolved in DMSO as a vehicle and the maximum volume used did not exceed 10 $\mu\text{l}/\text{ml}$ of media.

Cytotoxicity test

Following zeyleanol treatment, cell viability was assessed by the MTT (3-(4,5-dimethyl)-2,5-diphenyl tetrazolium bromide) assay¹³. This method is based on the ability of viable cells to reduce MTT and form a blue formazan product. MTT solution (sterile stock solution of 5 mg/ml) was added to cell suspension to a final concentration of 100 $\mu\text{g}/\text{ml}$ and the solution incubated for 4 h at 37°C in a humidified 5% CO_2 atmosphere. The medium was then removed and the cells were treated with DMSO for 30 min. The optical density (OD) of the cell lysate was measured at 540 nm, with a reference wavelength of 630 nm, using a microtitre plate reader (Biotek, USA). The number of viable cells was calculated from the number of untreated cells, and the data were expressed as percentage cell viability:

$$\text{cell survival} = \frac{\text{OD of sample}}{\text{OD of control}} \times 100\%.$$

Assay of caspase-3 activity

Cleavage of the fluorogenic peptide substrate DEVD-AMC (indicative of caspase-3-like enzyme activity) was estimated¹⁴. Cell lysates (1×10^6 cells) and substrate (50 mM) were combined in a standard reaction buffer and added to a 96-well plate. Enzyme-catalysed release of AMC was measured by a fluorescence plate reader (Bio-tek, USA) using 355 nm excitation and 460 nm emission wavelengths.

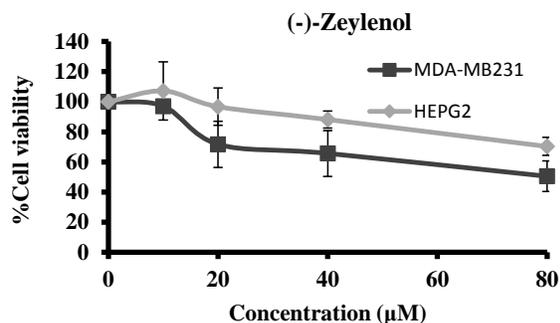


Fig. 2 Cytotoxic effects of (–)-zeaylenol on human breast cancer MDA-MB231, hepatocellular carcinoma HepG2 cell lines and MDA-MB231 or HepG2 cells were treated with the compound at various concentrations for 24 h and then the cell viability was determined by the MTT assay. The values were expressed as mean \pm S.D. from triplicates of three independent experiments.

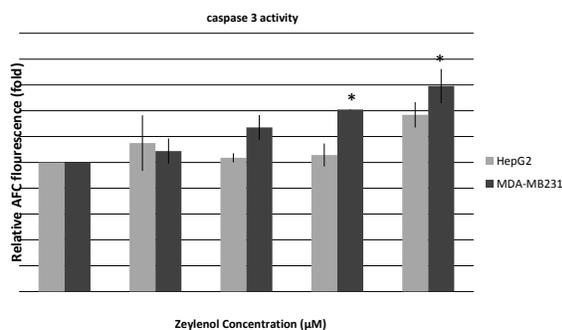


Fig. 3 Caspase-3 activity which was induced by (–)-zeaylenol in HepG2 and MDA-MB231 cells. HepG2 and MDA-MB231 cells were treated with the compound at various concentrations for 24 h and the caspase-3 activity was determined according to the method described in materials and methods. *, $p < 0.05$ as compared to control.

Statistical analysis

All data were expressed as mean \pm SE. Statistical comparison between groups was analysed by using one-way ANOVA and p values of less than 0.05 were considered significant.

RESULTS

(–)-Zeaylenol was evaluated for anti-inflammatory and cytotoxic activities. It was found that this compound showed similar anti-inflammatory activity as that observed for phenylbutazone (Table 1).

Moreover, (–)-zeaylenol exhibited moderate cytotoxic activity in human breast cancer MDA-MB231 at IC_{50} 54.22 μ M with SD 10.20 but with a lesser extent

in hepatocellular carcinoma HepG2 (Fig. 2). The caspase-3 activity, which increases in apoptotic cell death is a common effector proteolytic cysteine protease, increased significantly in MDA-MB231 cells (Fig. 3). This indicates cell death occur by apoptosis.

DISCUSSION

Anti-inflammatory effect

The ear oedema thickness of the rats in the control group increased gradually and peaked at 30 and 60 min after EPP application. The oedema persisted for 2 h during the assessment. Phenylbutazone, at a dose of 1 mg/ear, was used as a positive control, and it significantly reduced the oedema formation at all assessment times. The test zeaylenol significant inhibit oedema formation at the dose of 1 mg/ear at all determination times, with a similar intensity as phenylbutazone. The results of the present study reveal the anti-inflammatory activity of all test substance in the acute phase of inflammation. Formation of EPP-induced rat ear oedema is a useful model for screening and investigating the anti-inflammatory activity of test substances on the acute phase of inflammation. The inflammatory mediators released in this model include histamine, serotonin, bradykinin and prostaglandins. These mediators are capable of promoting vasodilation and increasing vascular permeability as well as synergistically producing oedema¹⁵. It was found that zeaylenol elicited a significant inhibitory effect on the oedema formation at all assessment times, at the dose of 1 mg/ear similar to that of phenylbutazone. Thus (–)-zeaylenol probably possesses anti-inflammatory activity by inhibition of the release or synthesis of various inflammatory mediators.

Anticancer effect

The (–)-zeaylenol was toxic to both MDA-MB231 and HepG2 cells in a dose response manner (Fig. 3) with IC_{50} values of 54 ± 10 and > 80 mM, respectively. Cell death occurred by apoptosis which is a programmed cell death indicated by the increase of caspase-3 activity a dose-dependent fashion in MDA-MB231 cell line at 40 and 80 μ M. In HepG2 cells, the caspase-3 activity was not significantly changed, which might indicate that the doses used were not cytotoxic to human hepatocellular cells. The maximum concentration to HepG2 cells was about IC_{30} . However, less than two fold increase in caspase-3 activity was observed in MDA-MB231 cell line. Furthermore, there was an overlap of error bars at 80 μ M for the two cell lines, indicating little difference in the effect of the compound between the two cell

Table 1 Anti-inflammatory activity of zeylenol.

Group	Dose (mg/ear)	oedema thickness (μm)				oedema inhibition (%)			
		15 min	30 min	1 h	2 h	15 min	30 min	1 h	2 h
control (acetone)	–	160.0 \pm 7.3	180.0 \pm 5.2	203.3 \pm 8.0	193.3 \pm 4.9	–	–	–	–
phenylbutazone	1	30 \pm 13*	43.3 \pm 9.5*	73 \pm 12*	90.0 \pm 4.5*	81	76	64	53
(–)-zeylenol	1	16.7 \pm 3.3*	57.7 \pm 2.1*	98.3 \pm 8.3*	93.3 \pm 5.6*	90	69	52	52

Results are expressed as mean \pm SE (number of ears = 6)

* Significantly different from the control group, $p < 0.05$

lines. (–)-Zeylenol might be used as an alternative medicine to treat cancer cells such as human breast cancer cells (MDA-MB231) or to lessen the dose of conventional chemotherapeutic drugs in combined treatment, which will reduce the toxic side effects of classical chemotherapy. However, experiments in vivo would need to be performed before the clinical application.

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